

Quantum Diffusion of Impurities in Solid Parahydrogen

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Over the last twenty years solid parahydrogen (pH₂) has emerged as an alternative host for matrix isolation studies in comparison to the more traditional noble gas solids. The physical and chemical properties of solid pH₂ distinguish it from noble gas solids in three important ways as a host for matrix isolation: (1) the weak cage effect of solid pH₂ allows *in situ* photolysis to be used as an effective means to produce isolated photoproducts, (2) several dopant species can readily diffuse in solid pH₂ permitting bimolecular solid-state reaction studies, and (3) the individual pH₂ host molecules can participate in chemical reactions while noble gas atoms are essentially chemically inert. This second distinction for a pH₂ host crystal has been leveraged by our research group¹ and others^{2,3} to study hydrogen atom (H atom) reactions with various chemical species in solid pH₂ where the H atom is mobile via a quantum diffusion process. Recently, we have observed the O(³P) + O₂(X ³Σ_g⁻) → O₃(X ¹A₁) reaction following the 193 nm *in situ* photolysis of O₂. This suggests that O(³P) atoms are mobile in solid pH₂ opening new opportunities to study O atom reactions. In fact, even stable molecules such as HF⁴ and H₂O⁵ have been reported to quantum diffuse in solid pH₂. We therefore have become interested in investigating which dopant species can diffuse, and which species are immobilized within solid pH₂. In this study we will focus on the diffusion properties of O(³P) and Cl(²P_{1/2}) atoms in solid pH₂.

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1. Mutunga, F.M.; Olenyik, K.M.; Strom, A.I.; Anderson, D.T. Hydrogen atom quantum diffusion in solid parahydrogen: The H+N₂O → *cis*-HNNO → *trans*-HNNO reaction. *J. Chem. Phys.* **2021**, *154*, 014302.
 2. Tsuge, M.; Tseng, C.-Y.; Lee, Y.-P. Spectroscopy of prospective interstellar ions and radicals isolated in solid *para*-hydrogen matrices. *Phys. Chem. Chem. Phys.* **2018**, *20*, 5344.
 3. Schneiker, A.; et al. Non-energetic, low-temperature formation of C_α-glycyl radical, a potential interstellar precursor of natural amino acids. *J. Phys. Chem. Lett.* **2021**, *12*, 6744.
 4. Ooe, H.; et al. Diffusion of hydrogen fluoride in solid parahydrogen. *J. Chem. Phys.* **2013**, *138*, 2104309.
 5. Moore, B.; Djuricanin, P.; Momose, T. Diffusion of water molecules in quantum crystals. *J. Phys. Chem. Lett.* **2018**, *9*, 6475.